CARBON DIOXIDE (CO2) IN THE ATMOSPHERE

New Zealand GHG

0.2 t/ha 0F CARBON BURPED AS METHANE. **EQUIVALENT TO** 5.7 t OF CO2

2.7 t/ha 0F CARBON RETURNED AS CO₂ THROUGH BREATH

10 t/ha 0F CARBON RETURNED AS CO2 THROUGH PLANT RESPIRATION

5 t/ha OF CARBON RELEASED IN THE DECAY OF PLANT **ROOTS AND** LITTER

1.6 t/ha OF CARBON EXPELLED IN WASTE

CARBON STORED IN FOSSIL FUELS 5 t/ha OF CO₂ IN **PASTURE** CONSUMED BY ANIMALS

CARBON IN SOILS

DID YOU KNOW?

GHG emissions come from

DID YOU KNOW?

figures given in this diagran

DID YOU KNOW?

New Zealand soils store 100-300 tonnes of

- GREENHOUSE GASES

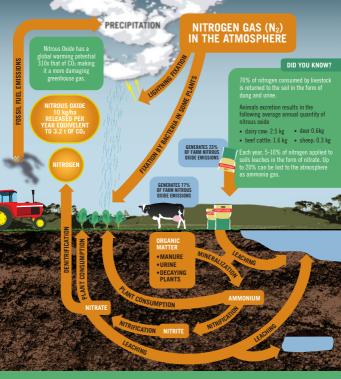
PASTURE AND

PLANTS TAKE 20 t/ha OF

CARBON AS CO2

FROM THE ATMOSPHERE. **EACH YEAR**

10 t/ha OF CARBON IS RETAINED IN PASTURE AND PLANTS



FIXATION

Where nitrogen gas is converted to ammonium by some plants and high-energy natural events such as lightning and forest fires.

MINERALISATION

When the decomposition of organic matter converts nitrogen to ammonium.

NITRIFICATION

Where ammonium in the soil is converted

into nitrite and then nitrate DENITRIFICATION

The conversion of nitrate into nitrogen or nitrous oxide gas.

LEACHING

Ammonium is attached to clay particles whereas nitrate is not. Therefore nitrate leaches readily.